













Microbicides & the environment

- \succ large quantities used in healthcare & industry
- importance of spent microbicide disposal for environment not yet widely recognized
- > all discarded to environment primarily water through sewage & land through sludge
- already concern over antimicrobials like antibiotics and trichlosan in drinking water
- > some microbicides used deliberately in water and sewage treatment



Disinfectant byproducts (DBP) > microbicides produce many DBP - high reactivity > DBP can be more toxic than original microbicide > only studied well for hypochlorite - now under study for other chlorine chemistries > regulated for water treatment > significant issue not widely considered outside of drinking water

- higher concentrations used in food production & discharged from processing
- paper production, sewage and many industries
- > need work on DBP for other microbicides 💽



- broadly used in water treatment, food sanitation and many industries
- hypochlorite & chloramines give chlorinated DBPs – many toxic, some mutagenic
- DBPs can be measured in breath of swimmers
 chlorine dioxide gives only oxidised DBP but needs on-site generation – not used in
- healthcare
 effective microbicides but readily neutralized and need careful use to ensure efficacy

Glutaraldehyde & OPA

- glutaraldehyde well recognized as sensitizer, respiratory irritant and cause of occupational asthma; less data on opa
- used at relatively high concentrations for instrument reprocessing and some exposure might be inevitable
- in europe also used for environmental surface disinfection – possibly more respiratory exposure

Quaternary ammonium compounds

- the most commonly used actives in microbicidal products
- act at membrane level pole holes in membranes and make them leaky
- relatively low human toxicity but still known to result in contact dermatitis and occupational asthma
- relatively refractory to environmental breakdown but can be used as carbon sources by variety of bacteria

Non-chlorine oxidisers

- hydrogen peroxide, peracetic acid, ozone
- chemicals often thought to be more environmentally benign than some microbicides because 'they do not leave a toxic residue'
- nevertheless oxidized DBPs will be present and not much is known about them
- extremely hazardous at high concentrations and if respired due to highly reactive nature

Quantitative structure activity relationships (QSARs) > toxicity of many chemicals remains untested

- Nature of the chemical often used to predict its toxicity from its structure and knowledge of the toxicity of related chemicals
- Used in human health and ecological risk assessment



Targets for toxicity unless swallowed, exposure usually insufficient to see acute organ effects effects more subtle and show in most sensitive systems at cellular level

- immune system/defence mechanism effects
- > genetic effects mutations
- > potential for carcinogenesis
- > potential for birth defects
- targets similar in humans and other species

Bacteria and toxins

- Fronically, bacteria often used to assess toxicity, mutagenicity of chemicals for humans, but almost no attention paid to the effects on bacteria
- e.g., if a bacterial test shows that a product is mutagenic, then it might be mutagenic for humans, but it is certainly mutagenic for bacteria
- bacterial-toxin interactions not generally seen as important for human health
- probably least explored interactions but may be very important
 - high surface area to volume ratio
 intimate contact & rapid reaction
 - nate contact & rapid reaction











Air pollution

- SO₂, NO₂, automobile fumes, ozone, many unknown chemicals
- \succ important particles in respirable range esp. <2.5 μm
- > small particulates, laden with chemicals and microbes, can pass directly into cells
- cellular & immune system effects
- might predispose to- or exacerbate infections
- asthma & atopy new cases or exacerbate symptoms?
- > chronic obstructive pulmonary disease (COPD)
- hypersensitivity pneumonitis
- increases parasitism of soil invertebrates by protozoa
 uncontaminated sites 0-20%, up to 80% at contaminated
- indoor air pollution (fungi, bacteria, endotoxins, chemicals)
 - > link between dampness, virus infection and allergen exposure
 - woodsmoke, tobacco smoke, virus infections and cance latent virus infection and cigarette smoke
- Water and food pollution
 air Scrubbing leads to similar spectrum of chemicals in water & foods
 effects of chemicals exacerbated by pathogen 'packages'
 simultaneous exposure to microbes and:

 pesticides and fertilizers food crops
 antibiotics, hormones and drugs food animals
 genotoxic contaminants in potable water include metals, low levels of pesticides, PCB's etc., disinfectant residuals, disinfection

byproducts

Metals and metalloids

- acute or chronic exposure may predispose to infections; augment or suppress immune response or make autoimmune. Examples:
- Mercury (induced autoimmunity and neurotoxicity) exacerbates virus infections and increases malaria in Hg-exposed
- Copper
 - b) reduced resistance catfish to Aeromonas hyrophila infection; rainbow trout to infectious hematopoietic necrosis virus and bacteria; Salmonids to Yersinia ruckeri (redmouth) – also viruses; Zebrafish – copper and zinc protective at low levels against Listeria monocytogenes, infection increase at higher levels Human infectione – elevated serum Cu in human
 - Human infections elevated serum Cu in human brucellosis, and in infertile men with Ureaplasma urealyticum (cause or effect?); Actinomyces israelii 2-12 % in Cu IUD users

Metals and metalloids cont.

- Cadmium (kidney, systemic toxin)
 heavy metal gradient (smelter) inc. infections at higher
 - metal (esp. cadmium) > Cd inc. stress response - scavenger enzymes protect
 - bacteria from host
 Single airborne Cd++ challenge increased mortality by Pasteurella multocida in mice but decreased it with Influenza A compared to Al+++ control challenge
- Cadmium increased *Listeria* infections in mice
 Zinc, Copper, Iron, selenium metabolism may be
- altered during infections
 Selenium needed in diet for proper immune function;
 Se deficiency increases viral pathologies in mice
- Arsenic greater mortality on challenge to
- streptococcal aerosol; reduced pulmonary bactericidal activity to *Klebsiella pneumoniae*



Microbial population changes

- microbicides & antibiotics kill more than targets
 kill all other susceptible bacteria that are carrying out useful and protective functions
- once the ecosystem is cleared of susceptible bacteria, resistant bacteria can multiply and dominate the environment due to lack of competition
- sometimes resistant bacteria are pathogens (e.g. mycobacteria)
- in general, microbial communities respond to presence of antimicrobial by shifts from those organisms that are sensitive to those that are tolerant or resistant
- sublethal exposure to microbicides can link to antibiotic resistance?







the ability of pathogens to rapidly evolve resistance to toxic chemicals in their environment gives them an unassailable advantage

even microbicides themselves are not impossible to colonize

their unique outer membrane that constitutes an effective barrier to the passage of germicides, and/or efflux systems. While the concentrated solutions of the disinfectants have not been demonstrated to be contaminated at the point of manufacture. Newmany et al. found that an unditude phenolic may be contaminated by a Pseudomonas sp. during use ¹¹⁸. The most of the reports that describe illness associated with contaminated disinfectants, they molt was used to disinfect patient-care equipment such as cystoscopes, cardiac catheters, and thermometers. The germoides used as disinfectants that were reported contaminated induce cholchexidine, quaternary amonium compounds, phenolic, and

"What does not kill me, makes me stronger".





- chemicals increasing, especially in healthcare settings > toxicology of many chemicals, & virulence factors of
- most microbes, only partially understood
 major gaps in knowledge of combined health impact of real-life exposures to chemicals & microbes
- > microbial control can create problems; microbicides useful but potentially dangerous – double edged sword
- > need prudent use for efficacy and safety; strategies for microbicide use to avoid sublethal exposures
- Are truly safe & effective biocides possible?
- > multidisciplinary work & sustained funding needed

Thank you for your attention "Soap and water and common sense are the best disinfectants" William Osler (1849-1919) Canadian physician

Bibliography

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